

22.02

INTRODUCTION TO
APPLIED NUCLEAR
PHYSICS

Lecture 2

SEMI-EMPIRICAL MASS FORMULA

$$M(Z, A) = Zm({}^1H) + Nm_n - B(Z, A)/c^2$$

● With binding energy given by:

$$B(A, Z) = a_v A - a_s A^{2/3} - a_c Z(Z - 1)A^{-1/3} - a_{sym} \frac{(A - 2Z)^2}{A} + \delta a_p A^{-3/4}$$



SEMI-EMPIRICAL MASS FORMULA

$$M(Z, A) = Zm({}^1H) + Nm_n - B(Z, A)/c^2$$

● With binding energy given by:

$$B(A, Z) = \underbrace{a_v A}_{\text{volume}} - \underbrace{a_s A^{2/3}}_{\text{surface}} - \underbrace{a_c Z(Z-1)A^{-1/3}}_{\text{Coulomb}} - \underbrace{a_{sym} \frac{(A-2Z)^2}{A}}_{\text{symmetry}} + \underbrace{\delta a_p A^{-3/4}}_{\text{pairing}}$$



CHART of NUCLIDES (Z vs. A)

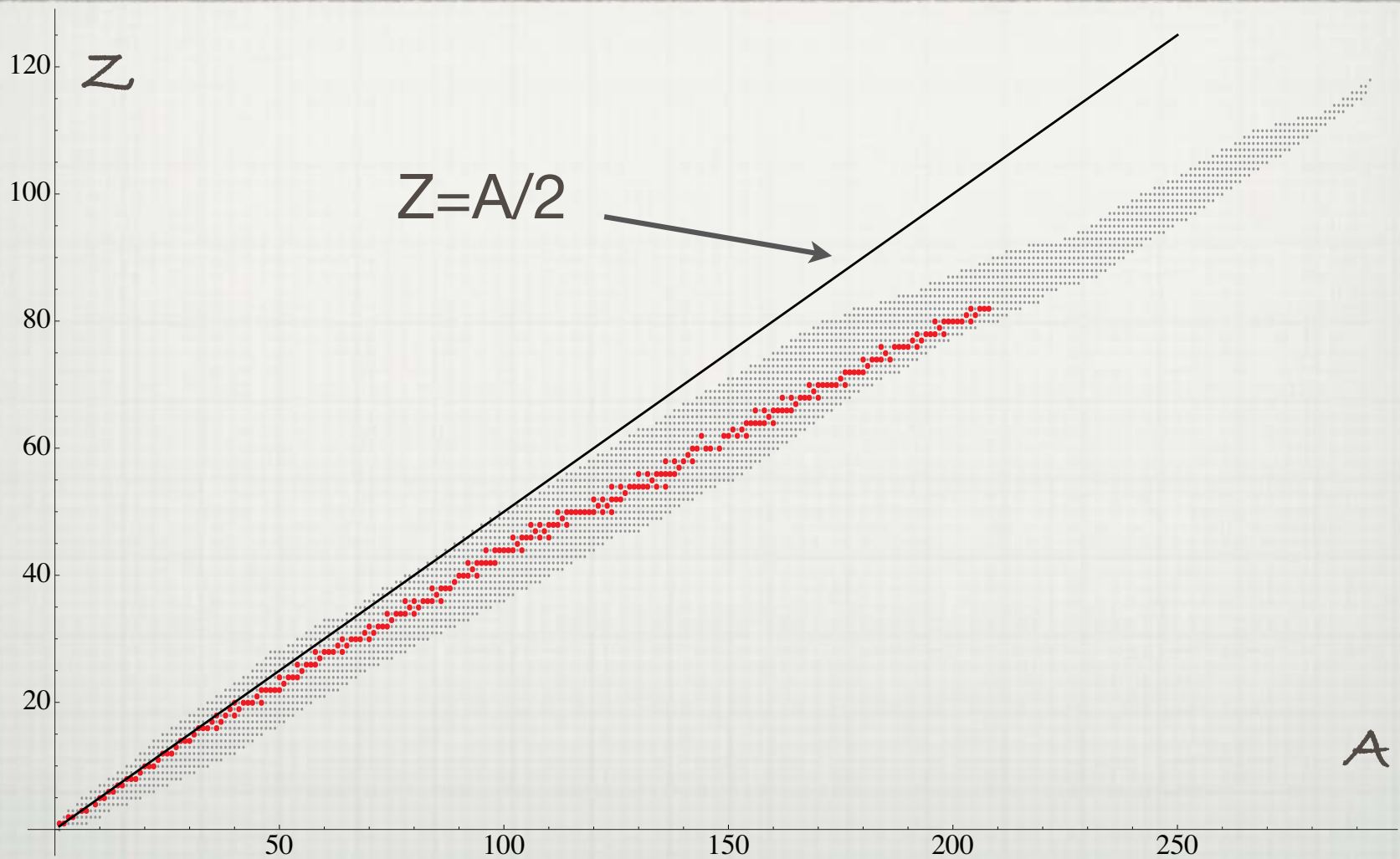
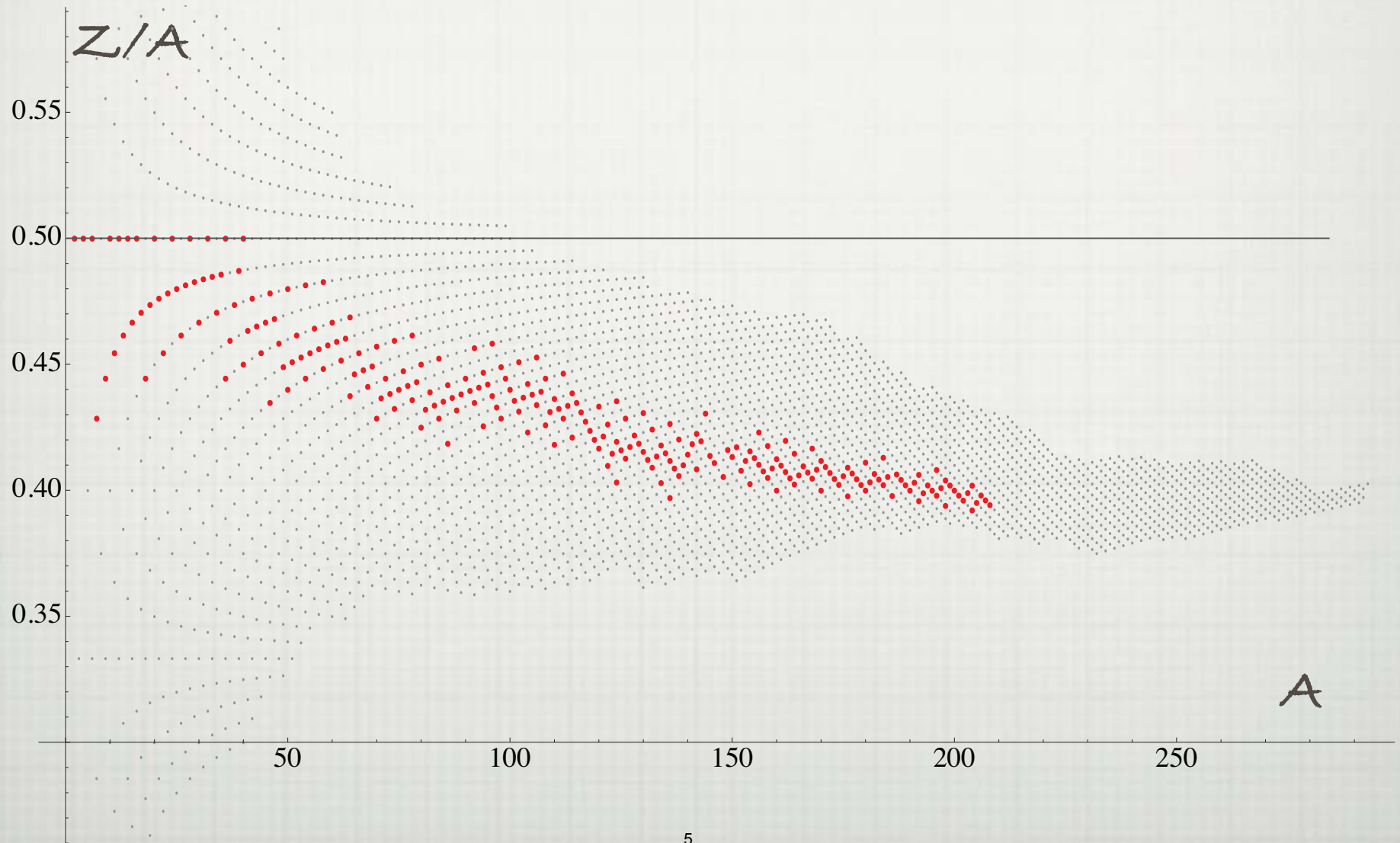
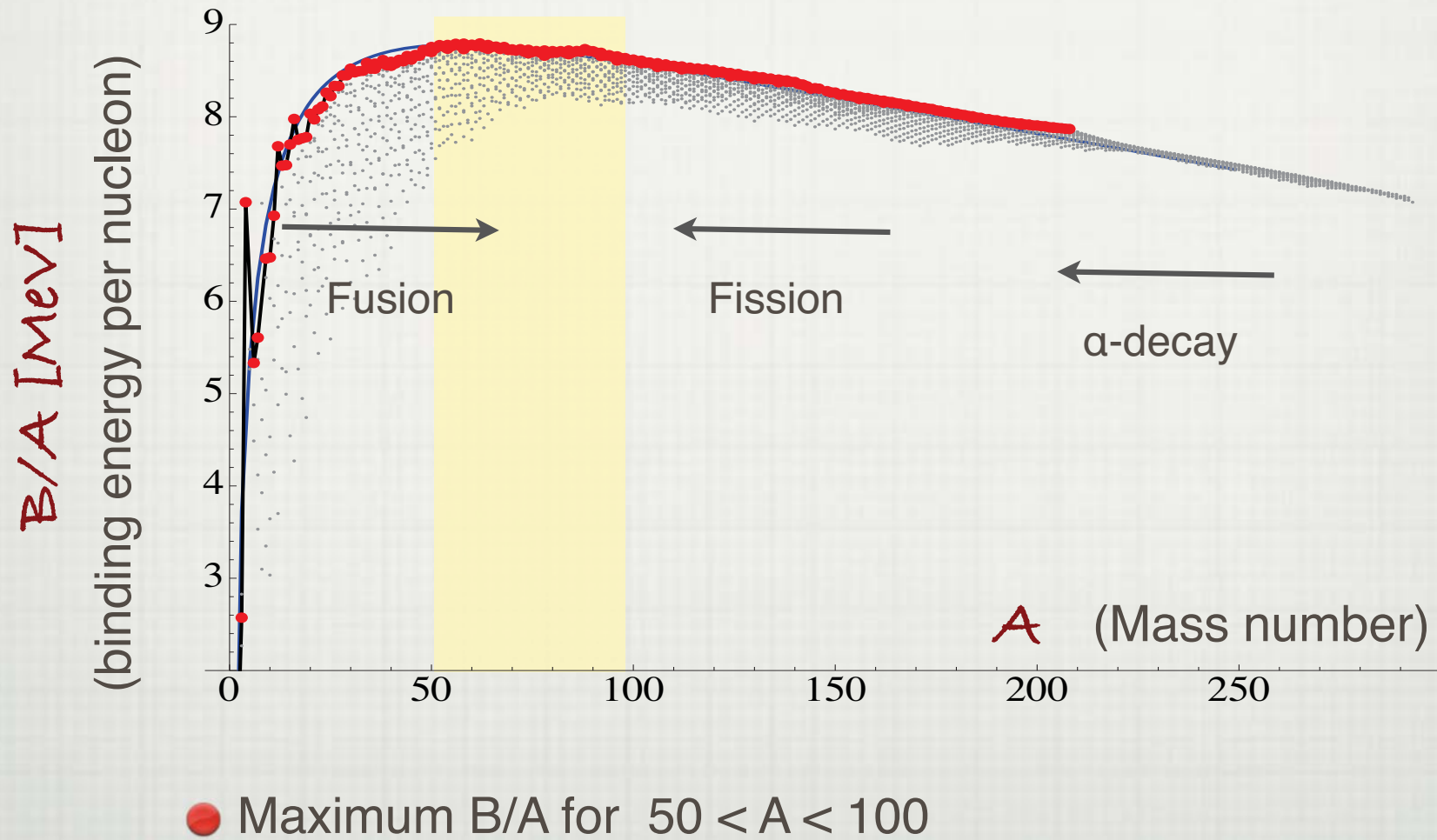


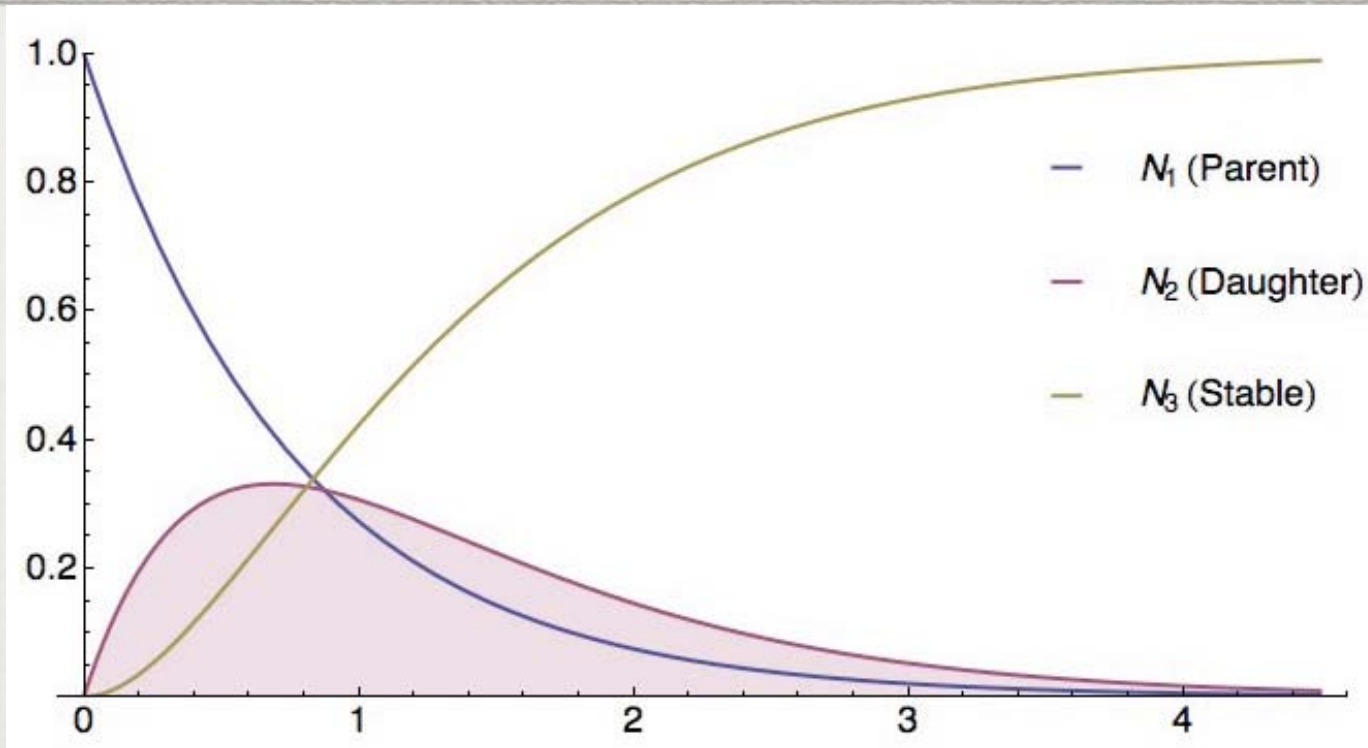
CHART of NUCLIDES (Z/A vs. A)



B/A: MAXIMUM



RADIOACTIVE DECAY CHAIN



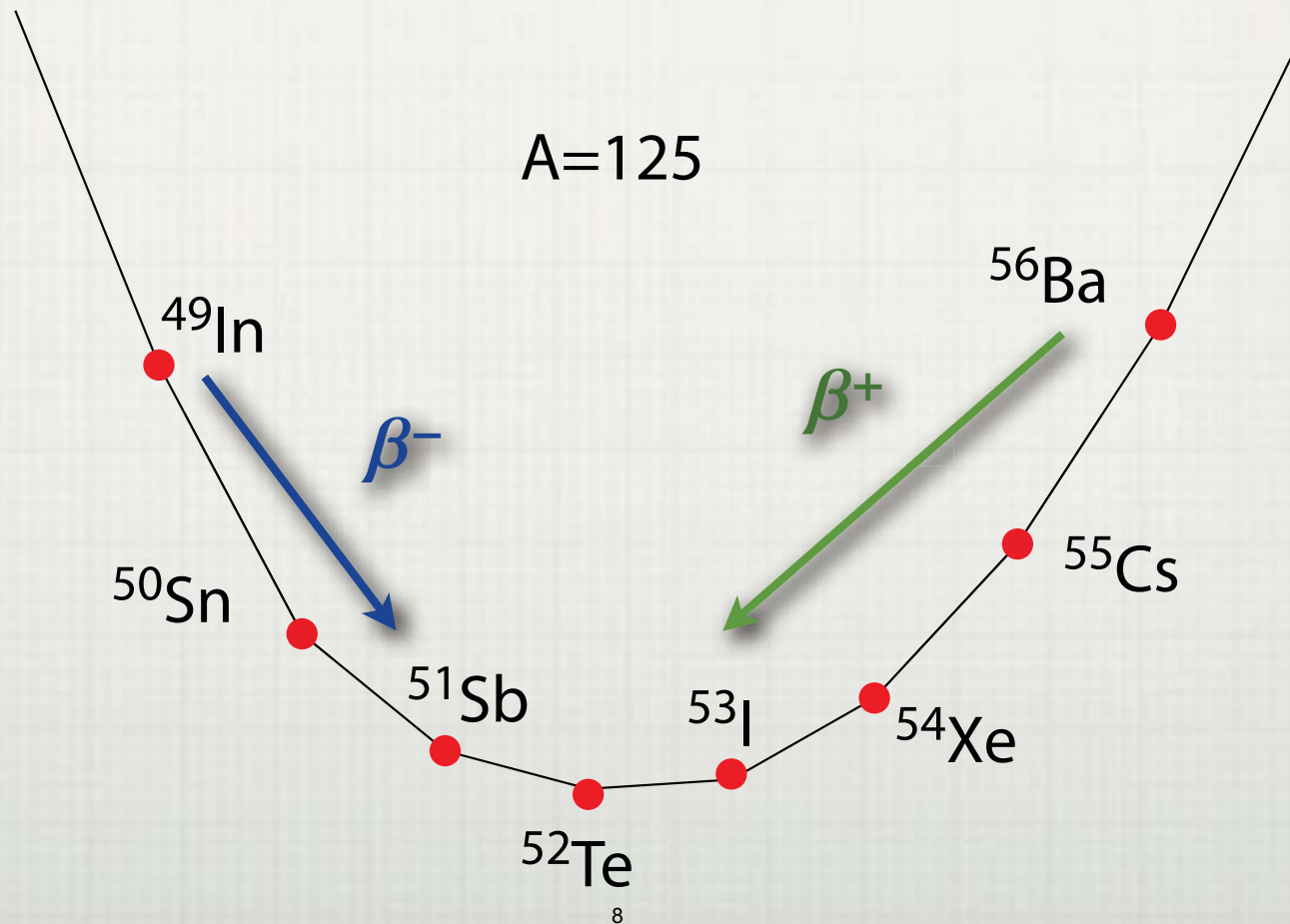
$$N_1(t) = N_0 e^{-\lambda_1 t}$$

$$N_2(t) = N_0 \frac{\lambda_1 (e^{-\lambda_1 t} - e^{-\lambda_2 t})}{\lambda_2 - \lambda_1}$$

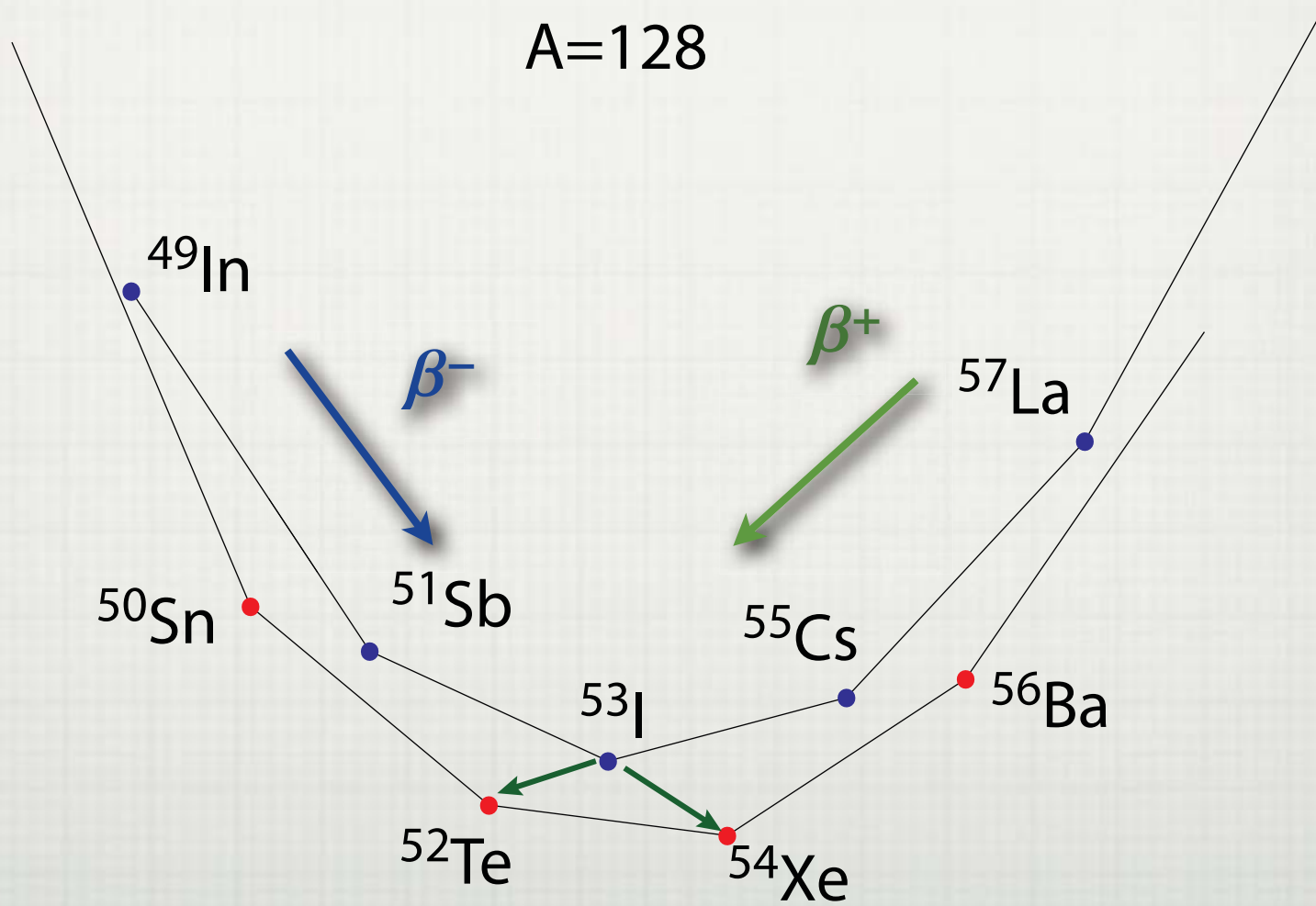
$$N_3(t) = N_0 \frac{\lambda_1 (1 - e^{-\lambda_2 t}) - \lambda_2 (1 - e^{-\lambda_1 t})}{\lambda_1 - \lambda_2}$$

MASS PARABOLA

- $M(Z, A = \text{cst})$: Mass of nuclides at constant mass # A



MASS PARABOLA



MASS PARABOLA

$M(Z, A = \text{cst})$: Mass of nuclides at constant mass # A

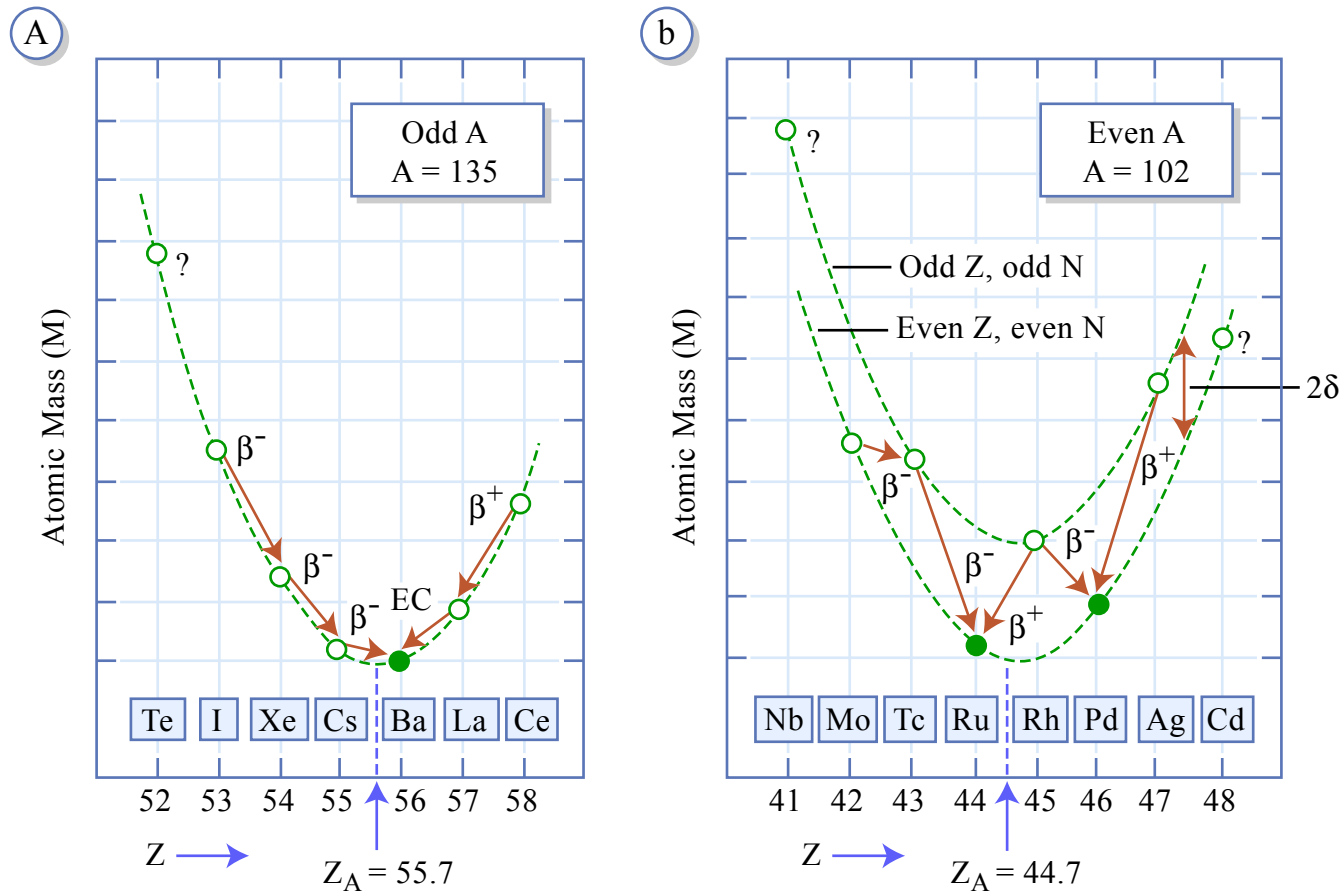


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