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2.61 Internal Combustion Engines
Spring 2008

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Assessment of Future Automotive Power Plant Technology

Fuel and engine alternatives

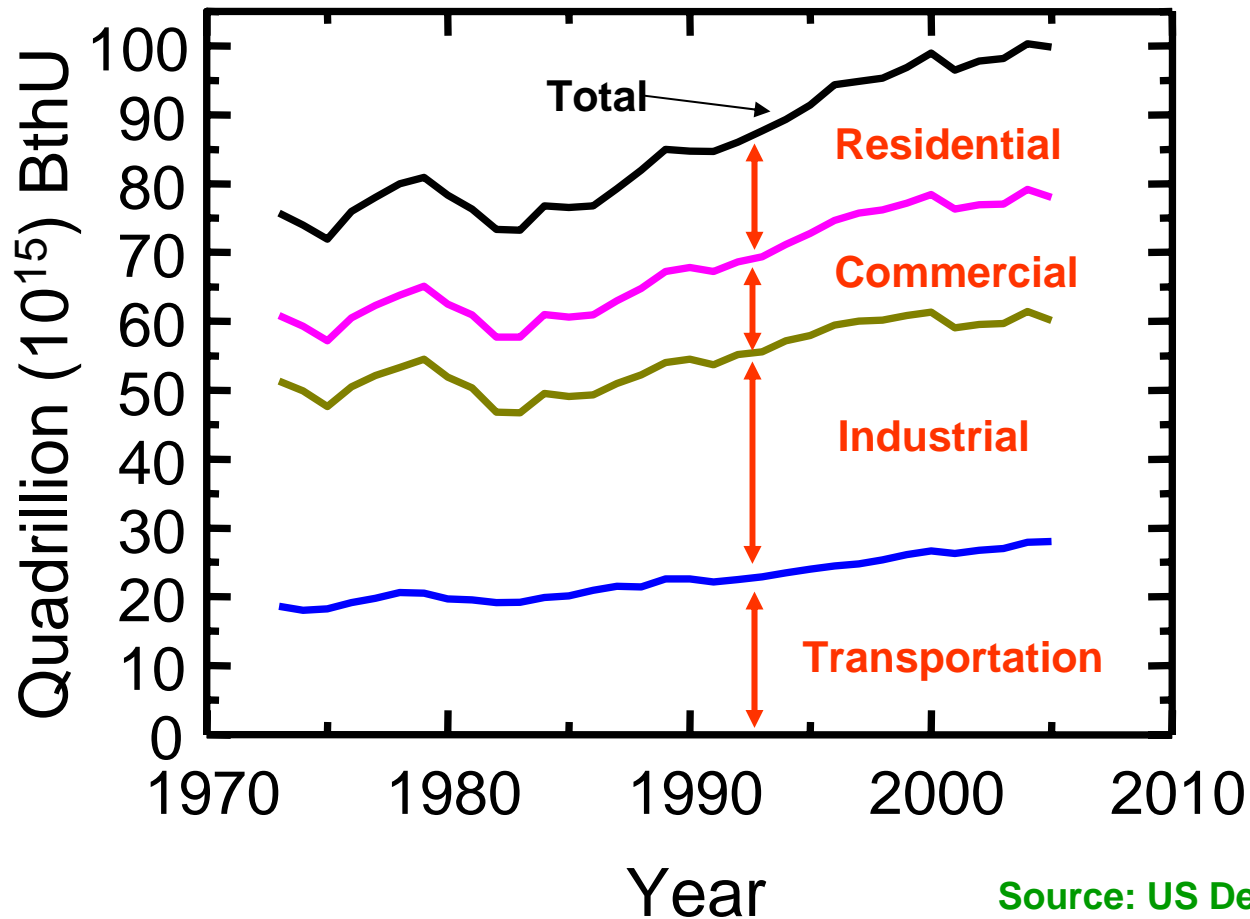
Prof. Wai K. Cheng
Sloan Automotive Lab
Massachusetts Institute of Technology

Transportation/Mobility

- Transportation/mobility is a vital to modern economy
 - Transport of People
 - Transport of goods and produce
- People get accustomed to the ability to travel

Transportation takes energy

US use of energy per year by sectors



Source: US Dept. of Energy

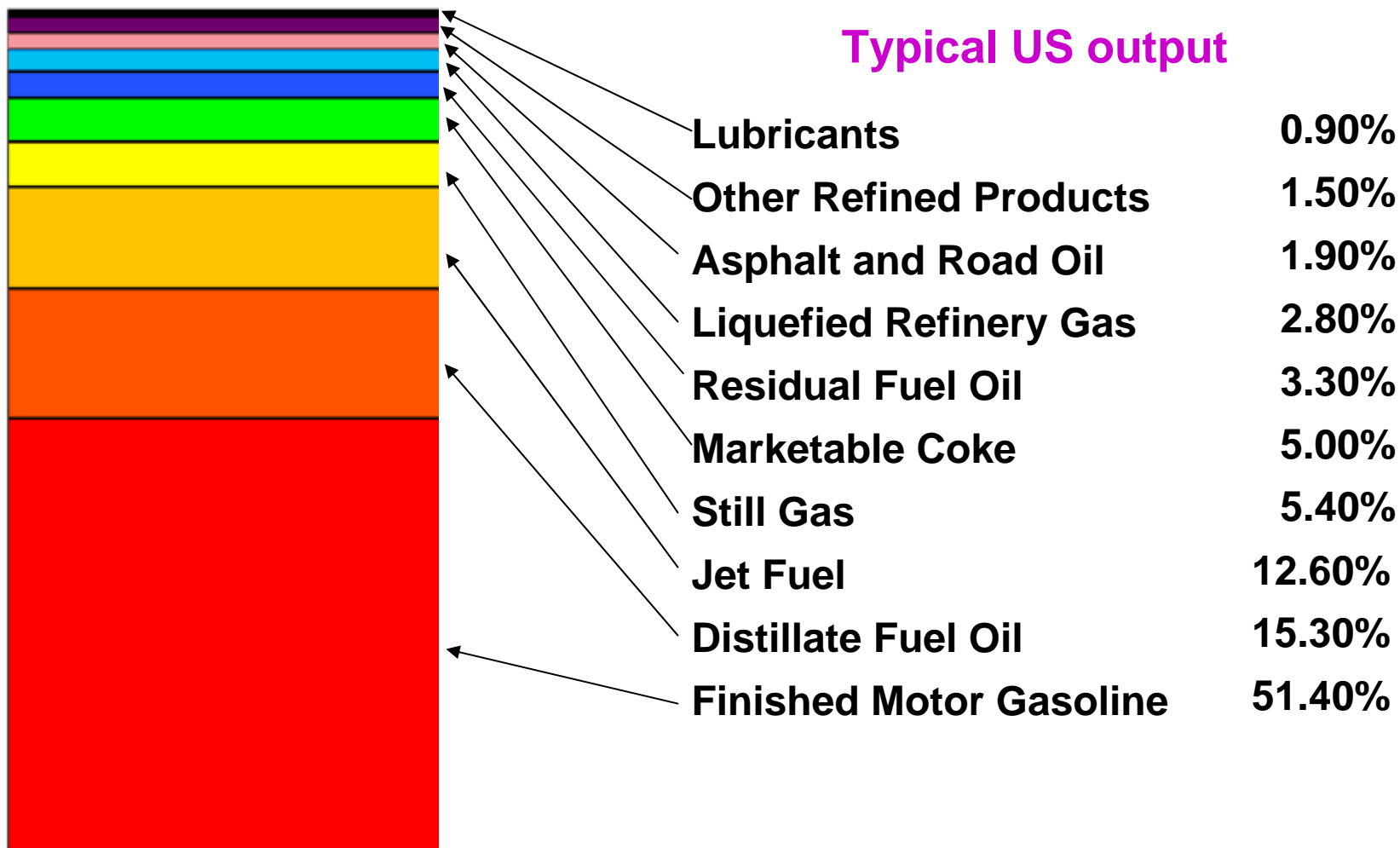
Transportation needs special kind of energy source

- Vehicles need to carry source of energy on board
- Hydrocarbons are unparalleled in terms of energy density
 - For example, look at refueling of gasoline
 - ~10 gallon in 2 minutes (~0.25 Kg/sec)
 - Corresponding energy flow
 - = 0.25 Kg/sec x 44 MJ/Kg
 - = 11 Mega Watts

Petroleum !

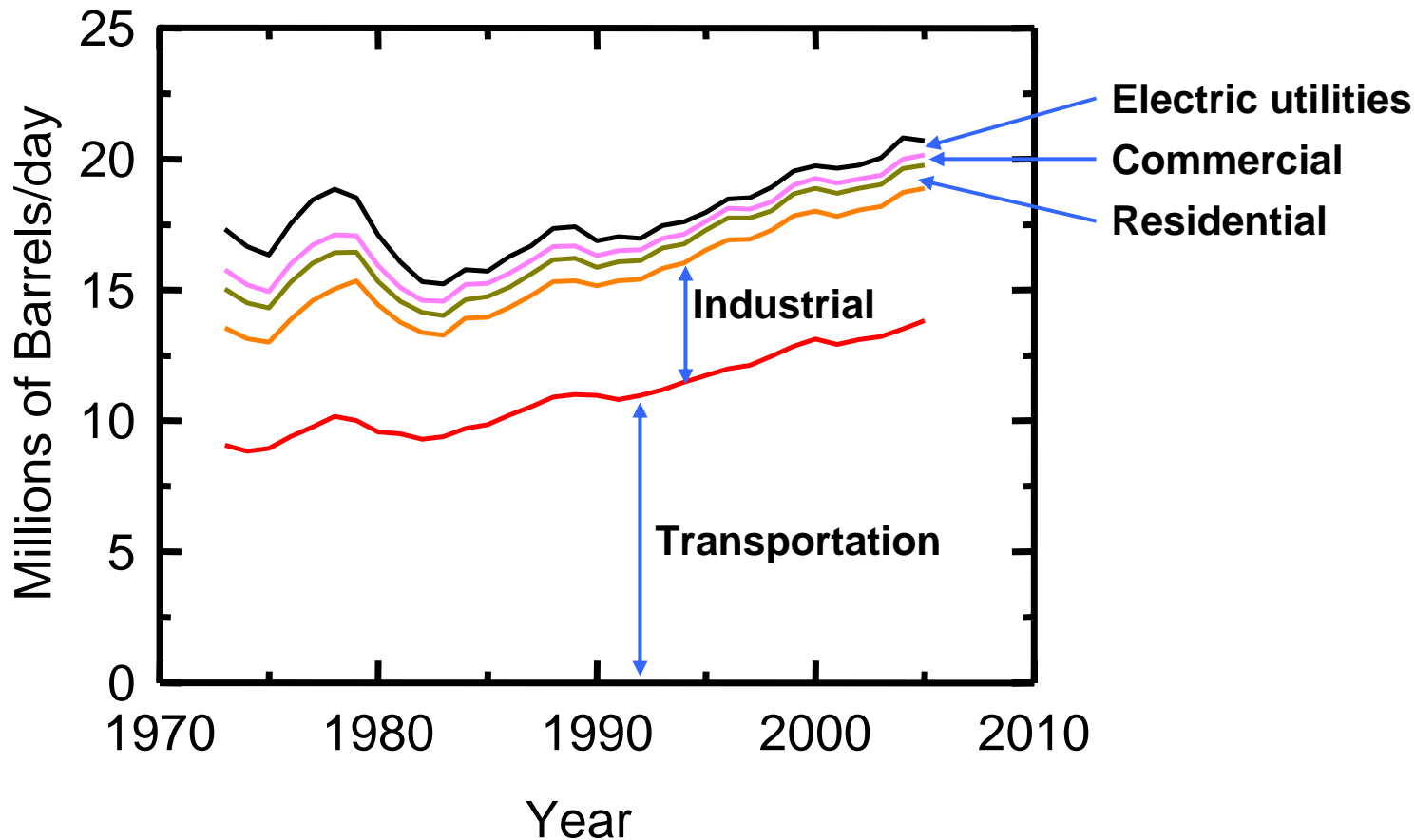
What is in a barrel of oil ?

(42 gallon oil → ~46 gallon products)



Source: California Energy Commission, Fuels Office

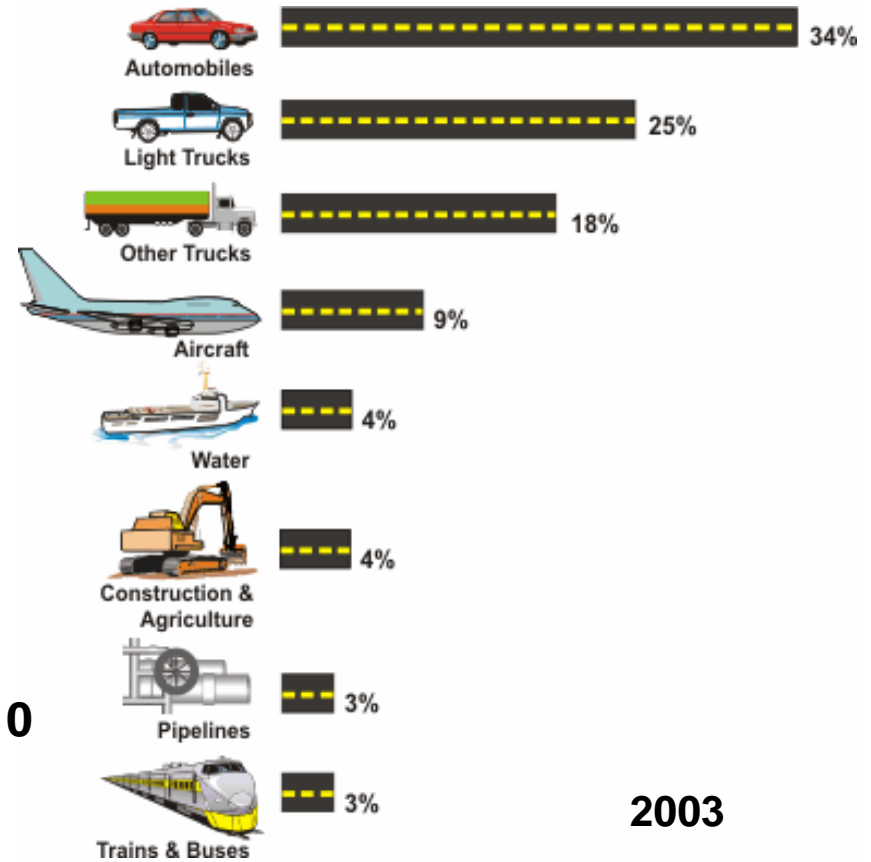
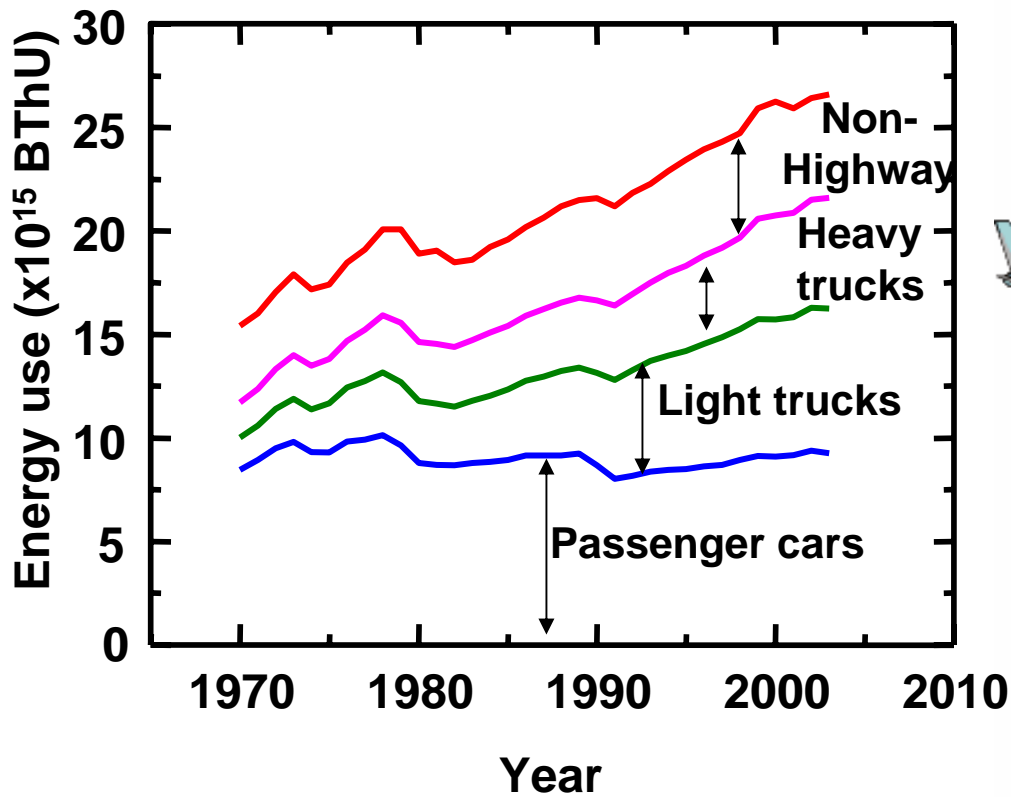
US Use of Petroleum by sector



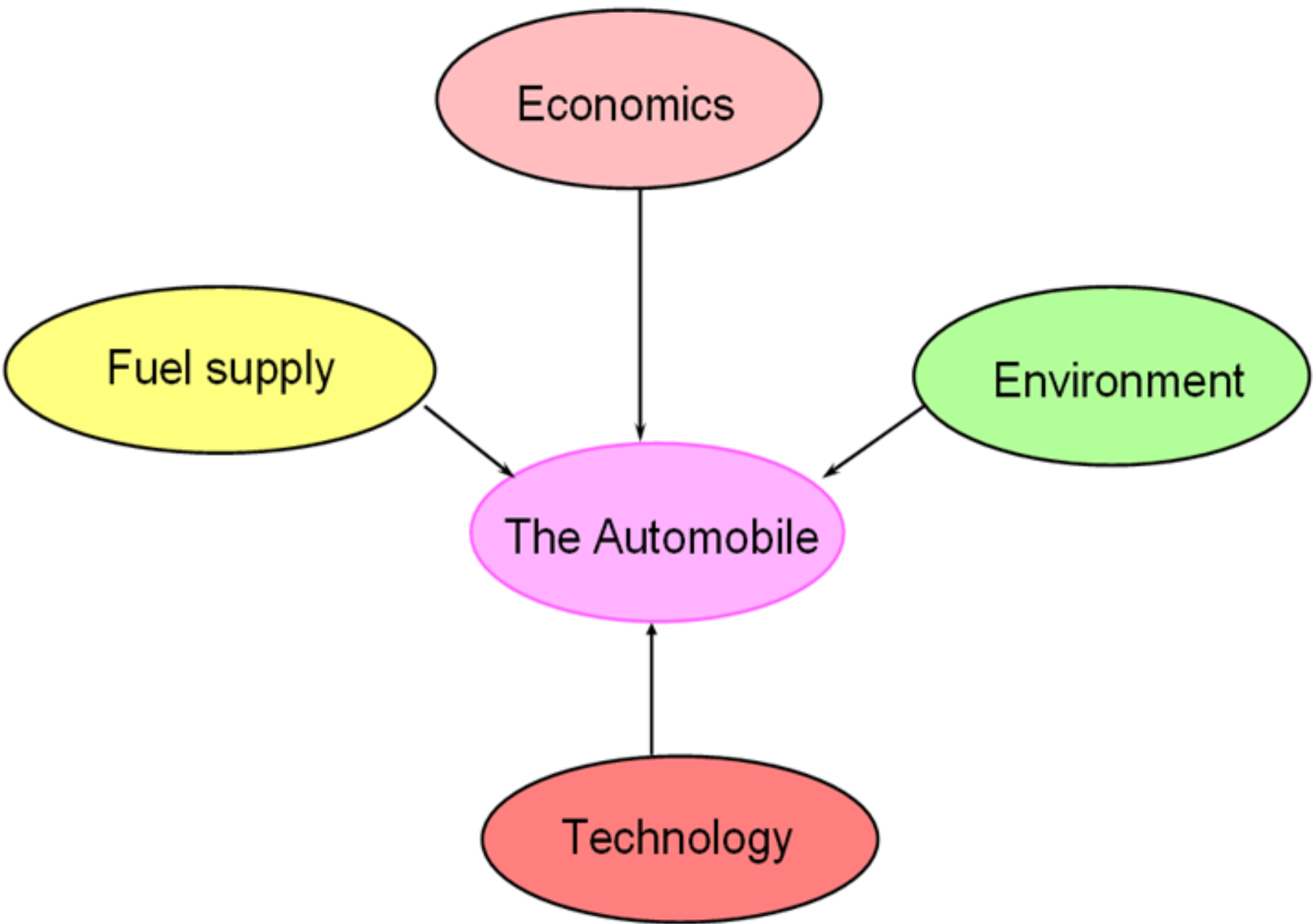
Source: US Dept. of Energy

Transportation energy use

(does not include military transportation)



Source: US Dept. of Energy, *Transportation Energy Data Book: Edition 26-2007*.



Size of the Automotive Industry

- Sales (US) ~ 18 millions new vehicles/year
- Approximately 72,000 vehicles produced per day (1.2 seconds / vehicle)



PRODUCT HAS TO BE ECONOMICALLY VIABLE ON ITS OWN

- High capital cost in manufacturing
- ~\$3 Billion or more for a new line



NEED HIGH VOLUME TO MAKE MONEY

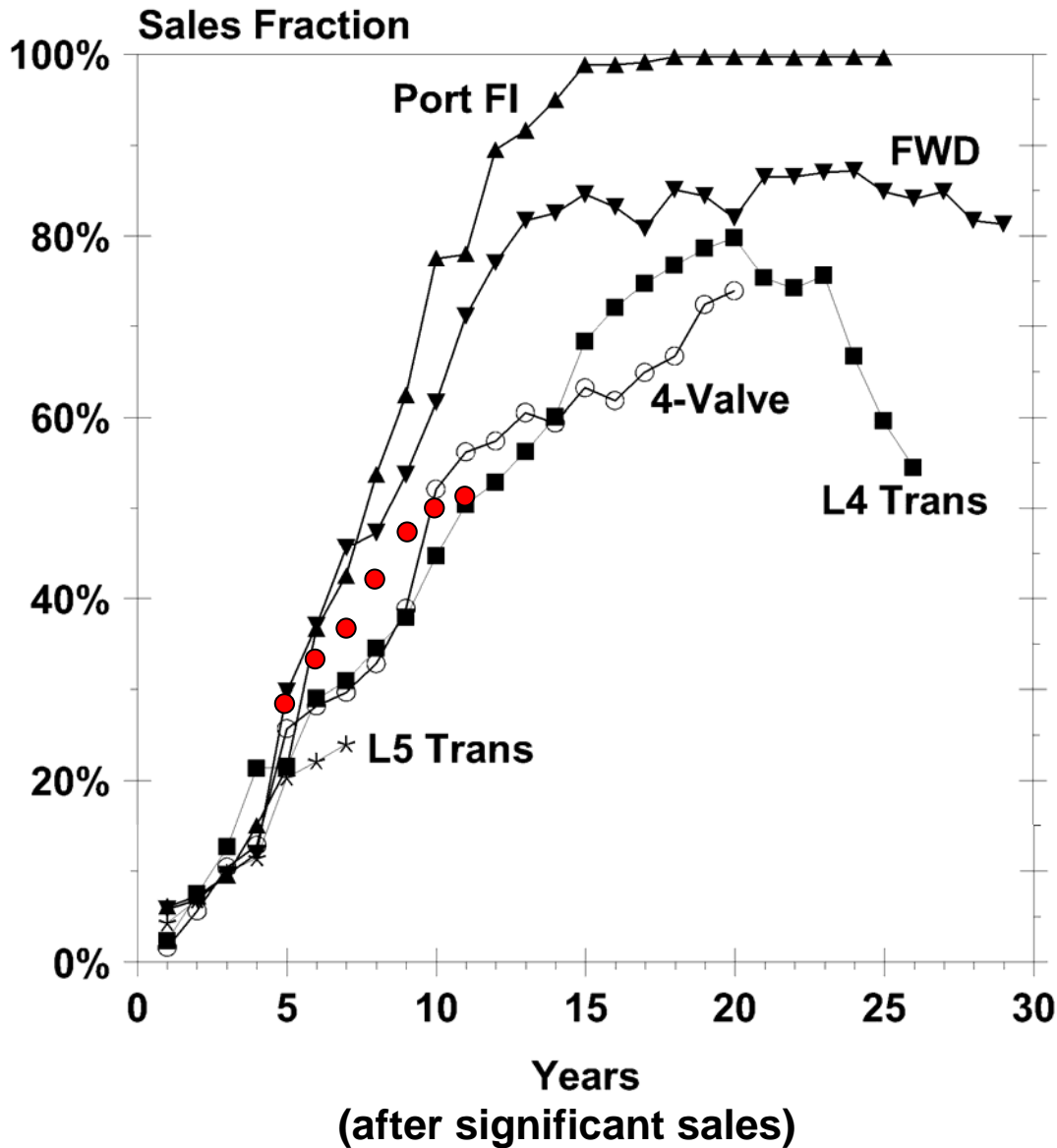
Petroleum Industry

Very capital intensive

- Exploration and production
- Refinery
- Distribution system

“Inertia” of the industry

- Utilization of capital
 - Need for capital expense to depreciate
- Technology takes time to develop and implemented
 - Example: vehicle powertrain
 - a. Incremental changes: Design needs to be completed 3-4 years before production
 - b. Significant changes: Add 5-10 years of development time to (a)
 - c. Drastic changes: Add 15 to 20 years to (a)
 - d. Radical changes: Add ? years to (a)
- Market penetration



Technology penetration

● Diesel sales fraction in Europe 1999-2005. (DI diesel introduced in 1997; sales fraction constant at 14% from 1987-1991.) Source: DOE

Source: Heavenrich, "Light-Duty Automotive Technology and Fuel Economy Trends, 1975-2005", EPA420-R-05-001

CUSTOMER NEEDS

VEHICLE

- Reasonable Cost
- Reliability
- Comfort
- Performance
- Aesthetics - Look and Feel

FUEL

- Cost
- Availability
- Ease of refueling

ENVIRONMENTAL IMPACT

- Air quality
 - NO_x
 - CO
 - Ozone
 - Particulate matters
 - Toxics
- Noise
- Green House Effect (CO₂, methane)
 - Kyoto Agreement (USA): 7% reduction of CO₂ from 1990 level
- Congestion

FUELS

- Reformulated Gasoline
- Methanol
- Ethanol and other bio-fuels
- Hydrogen

Transportation Fuels

Fuels	Density (Kg/m ³)	LHV/mass* (MJ/Kg)	LHV/Vol.** (MJ/m ³)	LHV/Vol. of Stoi.Mixture @1 atm, 300K*** (MJ/m ³)
Gasoline	750	44	3.3x10 ⁴	3.48
Diesel	810	42	3.4x10 ⁴	3.37
Natural Gas				
@1 bar	0.72	45	3.2x10 ¹ (x)	3.25
@100 bar	71		3.2x10 ³	
LNG (180K, 30bar)	270		1.22x10 ⁴	
Methanol	792	20	1.58x10 ⁴	3.19
Ethanol	785	26.9	2.11x10 ⁴	3.29
Hydrogen				
@1bar	0.082	120	0.984x10 ¹ (x)	2.86
@100 bar	8.2		0.984x10 ³	
Liquid (20K, 5 bar)	71		8.52x10 ³	

*Determines fuel mass to carry on vehicle

**Determines size of fuel tank

***Determines size of engine

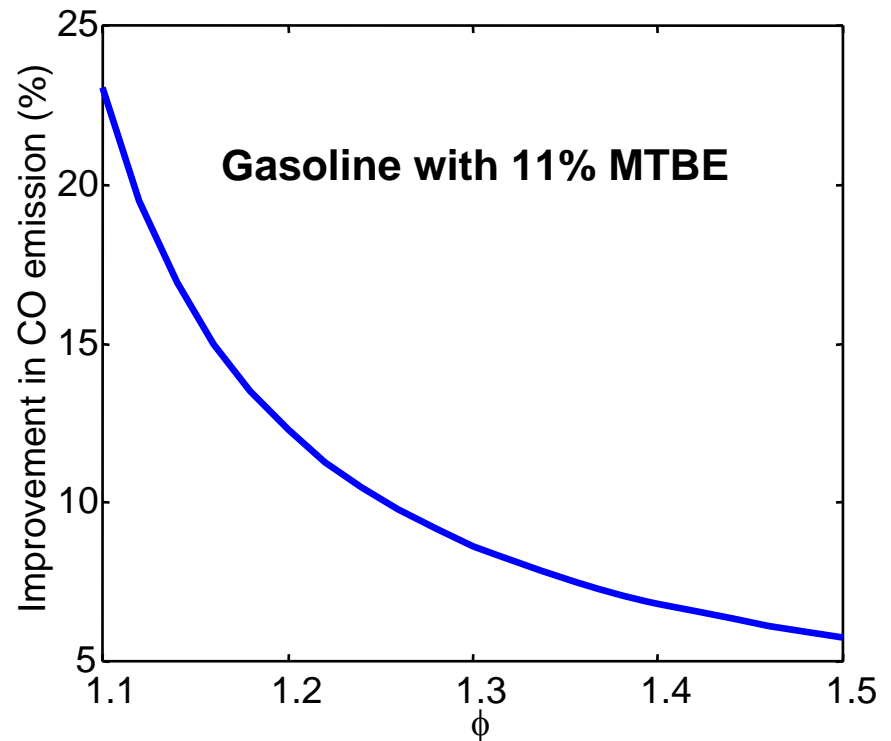
Relative CO₂ production from different fuel molecules

Image removed due to copyright restrictions. Please see Amann, Charles A. "The Passenger Car and The Greenhouse Effect." *SAE Journal of Passenger Cars* 99 (October 1990): 902099.

REFORMULATED FUELS

- Modify fuel properties to improve air quality (does not significantly impact CO₂ emissions)
- Introduce oxygenates (MTBE, ethanol, etc.) in gasoline to lower CO and HC emissions (US: 2% oxygenate required)
- Lower sulfur content
 - improve catalyst performance in gasoline vehicles
 - lowers sulfate emissions in diesels
- Lower aromatic content to reduce toxic emissions
- Lower Reid vapor pressure in gasoline to reduce diurnal emissions

- **COMPATIBLE WITH CURRENT ENGINES IN EXISTING FLEET**



Note: for modern engine with λ feedback, oxygenate effect on emissions is minimal

ALTERNATIVE FUEL: METHANOL

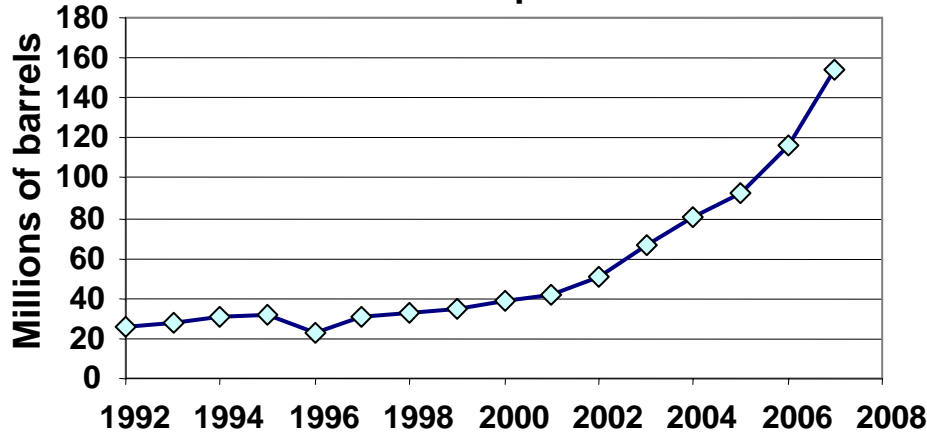
- GOOD COMBUSTION CHARACTERISTICS
 - High octane number (ON=99)
 - Cleaner exhaust: Lower CO and HC emissions
- PROBLEMS
 - Smaller heating value (~1/2 of that of gasoline)
 - toxic and corrosive
 - Difficulty in cold-start
- PRODUCTION - From natural gas and coal
 - Not efficient use of “original” fossil fuel: methanol is essentially a partially oxidized product
- **OUTLOOK**
 - Not an attractive intermediate alternatives because:
 - needs expensive retrofit of existing engine
 - Not good long term prospect; not efficient use of energy source

ALTERNATIVE FUEL: ETHANOL

- **GOOD COMBUSTION CHARACTERISTICS**
 - High octane number (ON=107)
 - Cleaner exhaust: Lower CO for older vehicles
- **PROBLEMS**
 - Smaller heating value (61% of that of gasoline)
 - Water absorption/corrosion/volatility problem
 - Need special hardware
 - Difficulty in cold-start
- **PRODUCTION**
 - Mostly from starch crops (corn, barley, wheat etc.) by fermenting and distilling
 - Cellulosic ethanol (from tree, grass, etc.)
- **E85 (85 liq. vol. % ethanol) is used as a practical fuel**
- **Needs flexible fuel vehicle for practical operation because of uncertainty in fuel supply**

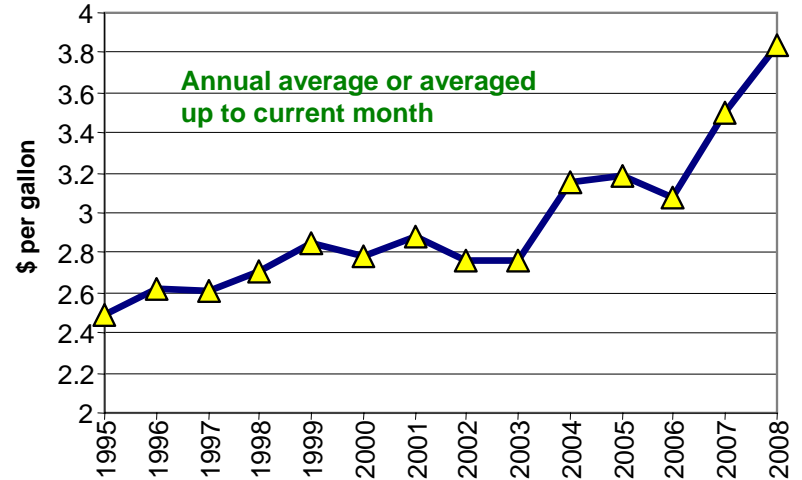
ALTERNATIVE FUEL: ETHANOL, bio-fuel for the future?

Annual fuel ethanol production

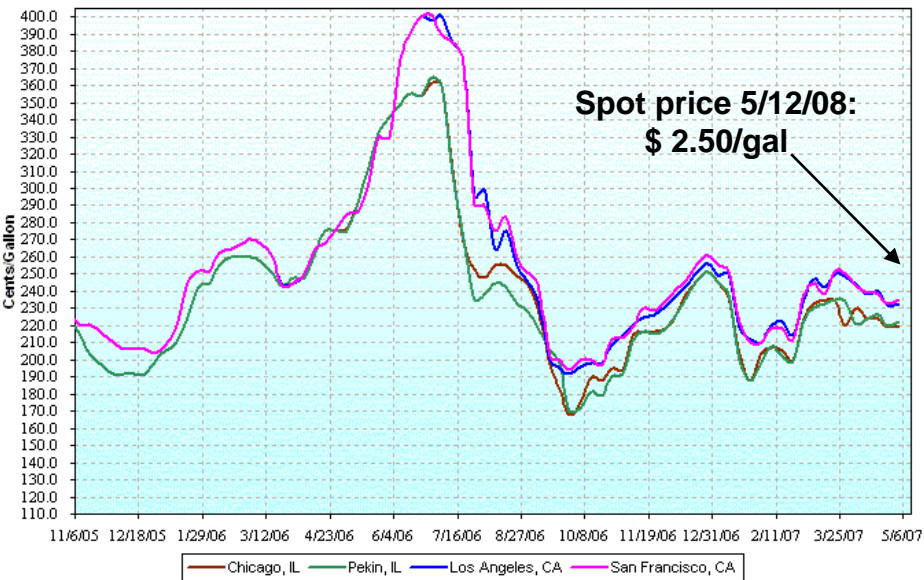


Source: California Energy Commission, 2006

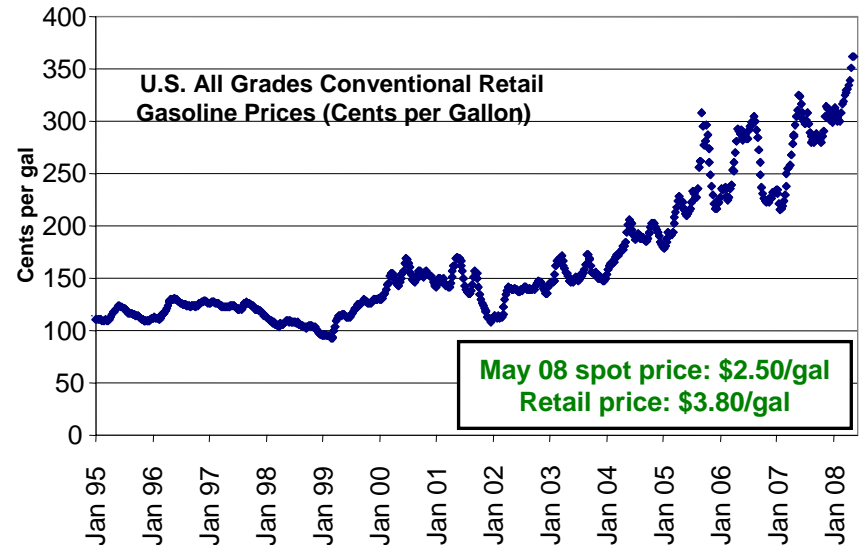
Fresh whole milk retail price (up to May, 08)



Fuel Ethanol Terminal Market Price - 18-Month History



Data Source: OXY-FUEL News Price Report. 1995-2005 Hart Publications, Inc.



May 08 spot price: \$2.50/gal
Retail price: \$3.80/gal

ALTERNATIVE FUEL: ETHANOL

Bio-Fuel for the future?

- Current US demand for ethanol is driven by government regulations and incentives
 - Ethanol flex-fuel vehicles produced because of the 74% credit towards CAFE requirement
 - (E85 vehicle equivalent mph = mpg x 1.74)
 - Gasoline oxygenate mandate, and phase out of MTBE
 - Energy bill (Aug. 05) mandated a threshold of 7.5 billion gallons (180 million barrels) production by 2012
 - Tax subsidy
 - blender's tax credit \$0.51/gallon alcohol
 - \$0.051/gallon fuel tax exemption for gasohol
 - minimum 10 vol % alcohol

Is corn-based ethanol the bio-fuel of the future?

➤ **Substantial increase in US food price**

ALTERNATIVE FUEL: ETHANOL

Bio-Fuel for the future?

Ethanol from corn

- Several studies of the overall energy budget

- P = energy used in production
 - feedstock production/ transport + processing
- E = Energy of the ethanol output
- Return (%) = $(E - P) / E$

- **Studies**

- Pimentel and Patzek (2003, 2005): **negative return**
 - Return = **-29%**
- USDA (Shapouri et al 2002, 2004): **positive return**
 - Return* = **+5.6%**
 - Return* = **+40%** if by products (Corn gluten meal, etc.) are accounted for

Verdict:
Substantial
environmental
and economic
cost; return not
clear

* For comparison purpose, these figures were converted from the values of $(E-P)/P$ of +5.9% and +67% in the original publication

Other bio-fuels

- Pimentel and Patzek also estimated energy budget for other bio-fuels. Returns:
 - Ethanol from switchgrass = -50%
 - Ethanol from wood biomass = -57%
 - Bio-diesel from soybean = -27%
 - Bio-diesel from sunflower = -118%
- Outlook: NOT CLEAR
 - New technology needed to change the picture

ALTERNATIVE FUEL: HYDROGEN

- Excellent fuel for combustion engines or fuel cells
 - No green house gas emissions/ hydrocarbon emissions
- Strictly, hydrogen is not a “fuel”, but an energy storage medium
- Not an efficient use of the “original” energy source
 - Efficiency loss in generating and in using the hydrogen
- PROBLEMS
- Storage (cryogenic, high pressure cylinders, metal hydride matrix) - Bulky and expensive
 - At 200 bar storage pressure, pumping loss is 13% of LHV
- Infra-structure for fuel supply
- Safety
- **OUTLOOK:** not attractive
 - On-board hydrogen storage: not a desirable option
 - Hydrogen from fuel reforming
 - Complex process with efficiency loss
 - Does not alleviate green house gas

ENGINES

- Spark Ignition Engines
 - Good fuel efficiency, reasonable cost
 - Improving emissions characteristics
- Diesel Engines
 - Better fuel economy
 - higher cost
 - NOx / particulate emissions
- Electric/ Hybrid/ Plug-in-hybrid Vehicles
- Fuel Cell

Hybrid vehicles

Configuration:

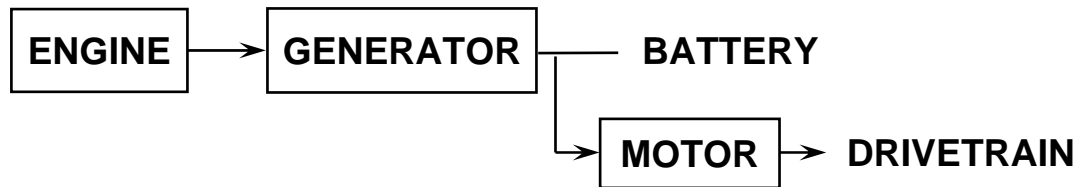
IC Engine + Generator + Battery + Electric Motor

Concept

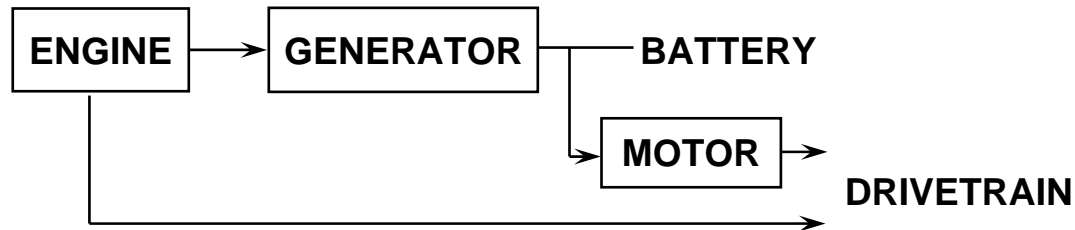
- Eliminates external charging
- As “load leveler”
 - Improved overall efficiency
- Regeneration ability
- Plug-in hybrids: use external electricity supply

Hybrid Vehicles

Series Hybrid



Parallel Hybrid



Examples: Toyota Prius (full hybrid); Honda Insight (electric assist)

Hybrid Vehicles: Market

- On the market since 1997 (Japan)
- Currently available in US:
 - Toyota Prius (~\$20K)
 - Honda Insight, Civic Hybrid (~\$19-20K)
 - Ford Escape (\$27K)
 - ...

Note:

No. of EV sold world wide since their introduction 30 years ago is < 30,000 units, and has flattened out

No. of Prius sold in three years(1997-2000)

➤ 34,000 units

Toyota Hybrid sale (2004) 130,000 units

(source: Toyota)

Toyota Prius

Honda Insight

Photos removed due to copyright restrictions. Please see any promotional photos of the Toyota Prius and the Honda Insight.

66/43 mpg on
Japan/US driving cycle

80/60 mpg on
Japan/US driving cycle

ELECTRIC HYBRID VEHICLE TECHNOLOGY

Toyota Prius

- Engine: 1.5 L, Variable Valve Timing, Miller Cycle (13.5 expansion ratio), Continuously Variable Transmission
 - 58 HP at 4000 rpm
- Motor - 40 HP
- Battery - Nickel-Metal Hydride, 288V
- Fuel efficiency:
 - 66 mpg (Japanese cycle)
 - 43 mpg (EPA city driving cycle)
 - 41 mpg (EPA highway driving cycle)
- Efficiency improvement (in Japanese cycle) attributed to:
 - 50% load distribution; 25% regeneration; 25% stop and go
- Cost: ~\$20K (subsidized)

Cost factor

If $\Delta\$$ is price premium for hybrid vehicle
 P is price of gasoline (per gallon)
 δ is fractional improvement in mpg

Then mileage (M) to be driven to break even is

$$M = \frac{\Delta\$ \times \text{mpg}}{P \times \delta}$$

(assume that interest rate is zero)

Cost Factor

Example:

Honda Civic and Civic-Hybrid

Price premium (Δ \$, MY08 listed)	= \$7155	<small>(\$22600-15445)</small>
mpg (city and highway av.)	= 29 mpg	(42 for hybrid)
hybrid improvement in mpg(%)	= 45%	

At gasoline price of \$4.00 per gallon, mileage (M) driven to break even is

$$M = \frac{\$7155 \times 29}{\$4 \times 45\%} = 115,000 \text{ miles}$$

(excluding interest cost)

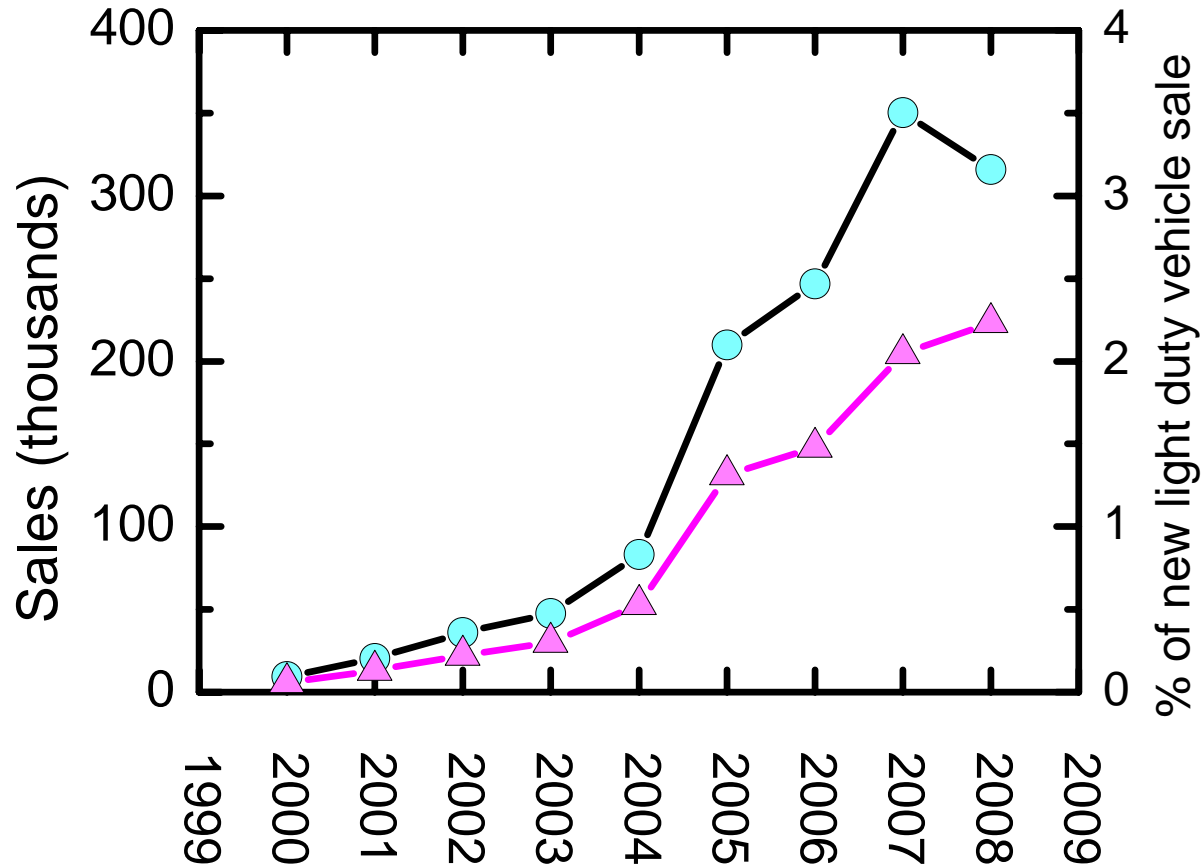
Barrier to Hybrid Vehicles

- Cost factor
 - difficult to justify especially for the small, already fuel efficient vehicles
- Battery replacement (not included in the previous breakeven analysis)
 - California ZEV mandate, battery packs must be warranted for 15 years or 150,000 miles : a technical challenge

Hybrid Vehicle Outlook

- Hybrid configuration will capture a fraction of the passenger market, especially when there is significant fuel price increase
- Competition
 - Customers downsize their cars
 - Small diesel vehicles
- Plug-in hybrids?
 - Weight penalty (battery + motor + engine)
 - No substantial advantage for overall CO₂ emissions
 - Limited battery life

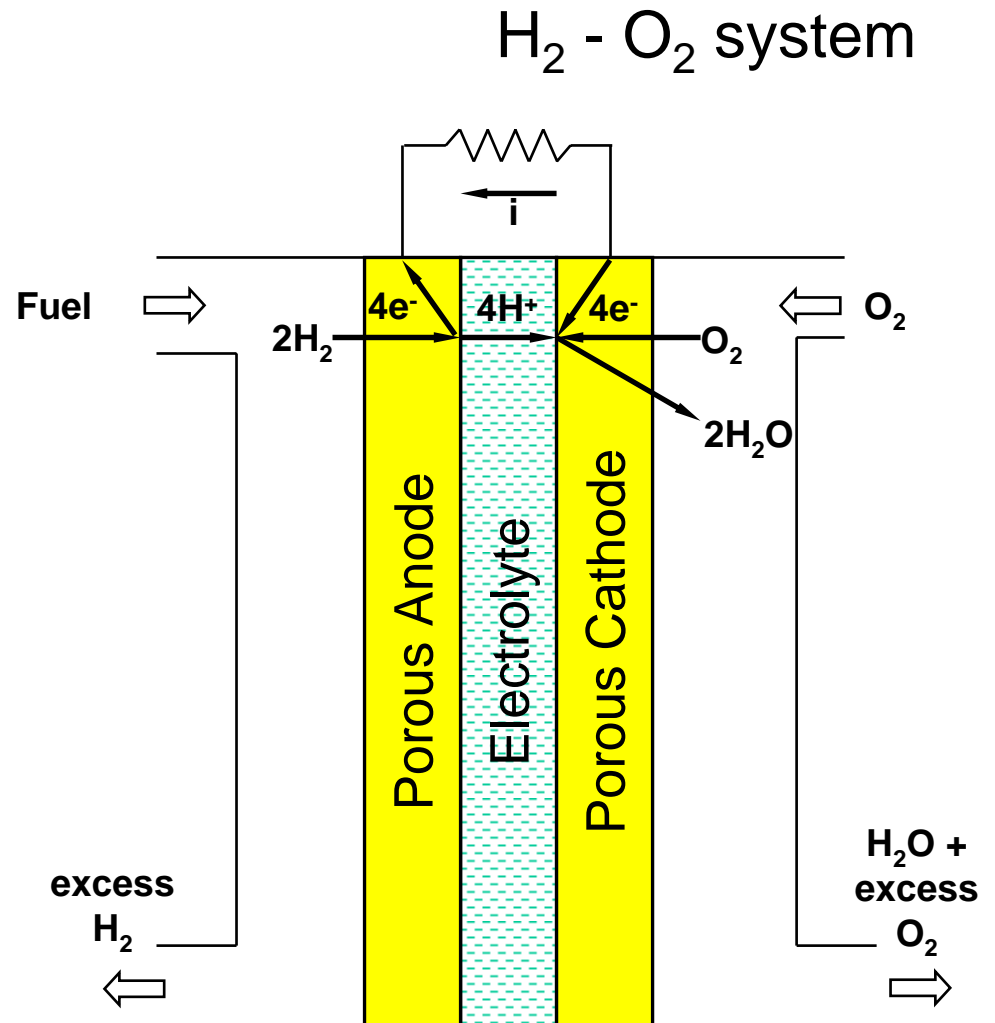
Sales figure for hybrid vehicles



What is a fuel cell?

Direct conversion
of fuel/oxidant to
electricity

- $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
- Potentially much higher efficiency than IC engines



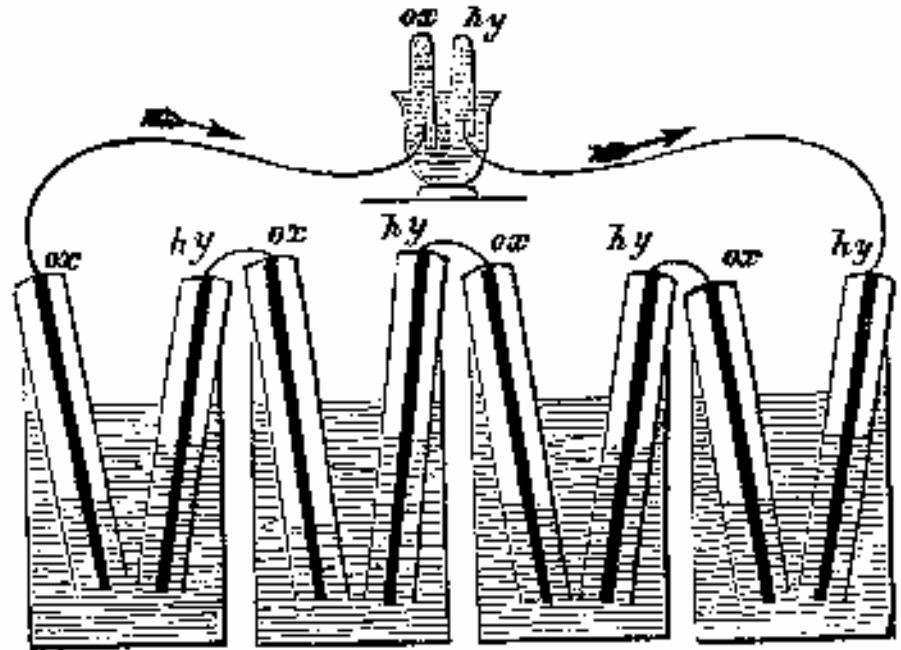
History of Fuel Cell

- **Sir William Grove** demonstrated the first fuel cell in 1839 (H₂ – O₂ system)
- Substantial activities in the late 1800's and early 1900's
 - Theoretically basis established
 - Nerst, Haber, Ostwald and others
- Development of Ion Exchange Membrane for application in the Gemini spacecraft in the 1950/1960
 - W.T. Grubb (US Patent 2,913,511, 1959)
- Development of fuel cell for automotive use (1960's to present)

The Grove Cell (1839)

- Important insights to fuel cell operation

- H₂-O₂ system (the most efficient and the only practical system so far)
- Platinum electrodes (role of catalyst)
- recognize the importance of the coexistence of reactants, electrodes and electrolyte



W.R.Grove, 'On Gaseous Voltaic Battery,' *Pil. Mag.*, **21**,3,1842

As appeared in Liebhafsky and Cairns, *Fuel Cells and Fuel Batteries*, Wiley, 1968

Types of fuel cell

- Classification by fuel
 - Direct conversion
 - Hydrogen/air (pre-dominant)
 - Methanol/air (under development; electrode poisoning problem)
 - Indirect conversion
 - reform hydrocarbon fuels to hydrogen first
- Classification by charge carrier in electrolyte
 - H^+ , O^{2-} (important difference in terms of product disposal)

Types of fuel cell (cont.)

- By electrolyte
 - Solid oxides: ~1000°C
 - Carbonates: ~600°C
 - H_3PO_4 : ~200°C
 - Proton Exchange Membrane (PEM): ~80°C

Automotive application

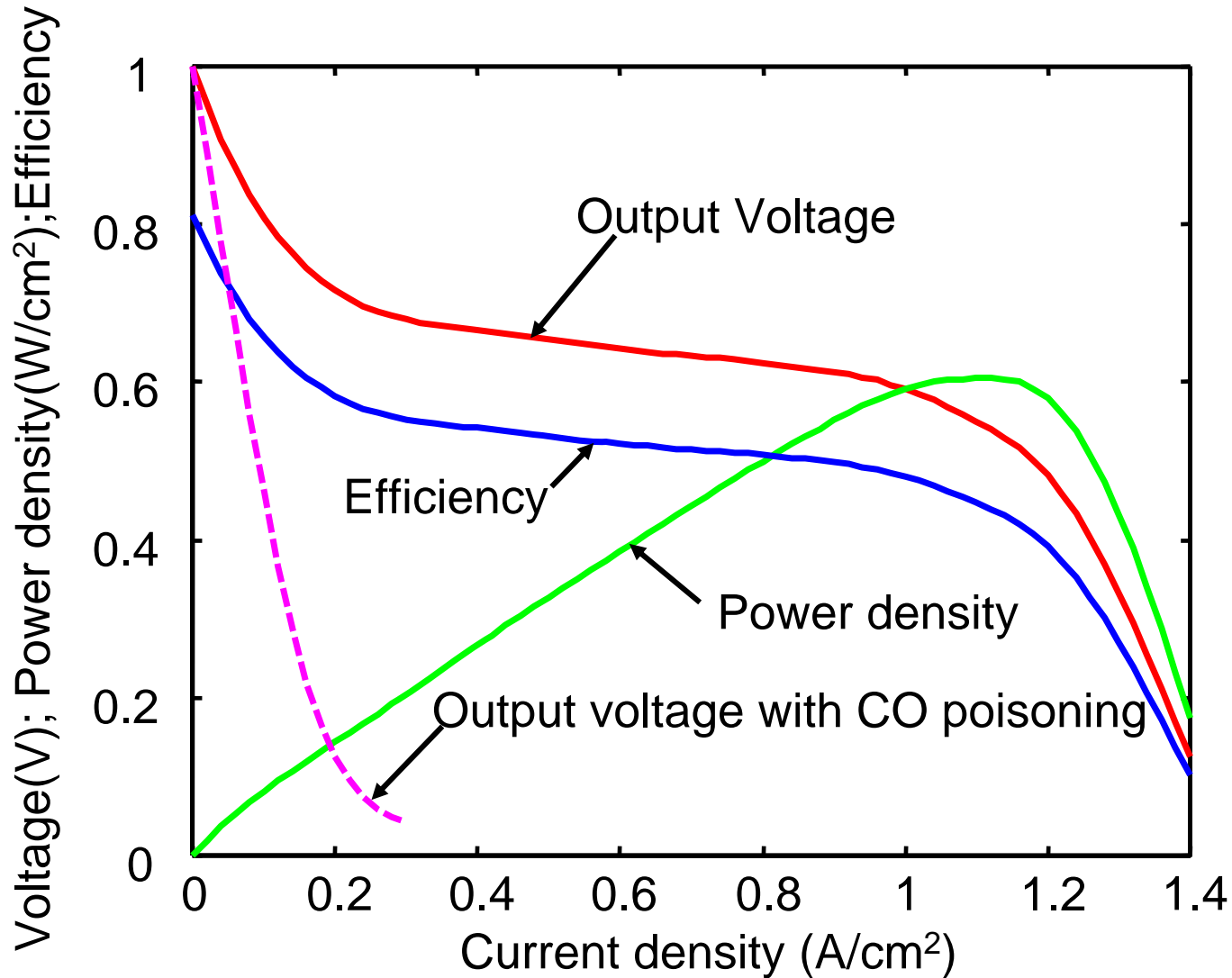


Modern PEM fuel cell stack

Diagram of a PEM fuel cell stack removed due to copyright restrictions.
Please see <http://www.technopr.com/download/Figure1-FuelCellConstruction.jpg>

(From 3M web site)

Current PEM H₂/O₂ Fuel Cell Performance



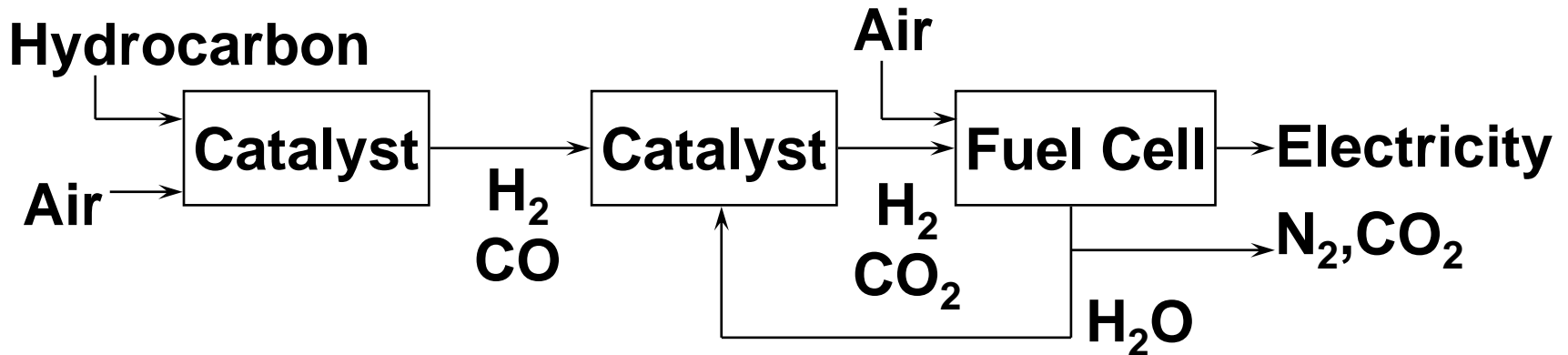
Note: Efficiency does not include power required to run supporting system

The Hydrogen problem:

Fundamentally H₂ is the only feasible fuel for fuel cell in the foreseeable future

- Strictly, hydrogen is not a “fuel”, but an energy storage medium
 - Difficulty in hydrogen storage
 - Difficulty in hydrogen supply infra structure
- Hydrogen from fossil fuel is NOT an efficient energy option
- Environmental resistance for nuclear and hydroelectric options

The hydrogen problem: H₂ from reforming petroleum fuel



Note: HC to H₂/CO process is exothermic;
energy loss ~20% and needs to cool stream
(Methanol reforming process is energy neutral, but
energy loss is similar when it is made from fossil fuel)

Current best reformer efficiency is ~70%

Problems:

CO poisoning of anode

Sulfur poisoning

Anode poisoning requires S<1ppm

Reformer catalyst poisoning requires S<50ppb

Fuel cell powerplant with fuel reforming

Images removed due to copyright restrictions. Please see photos of the Chevrolet S-10 Gen III gasoline fuel cell vehicle, such as <http://www.pickuptrucks.com/html/news/fuelcells10.html>

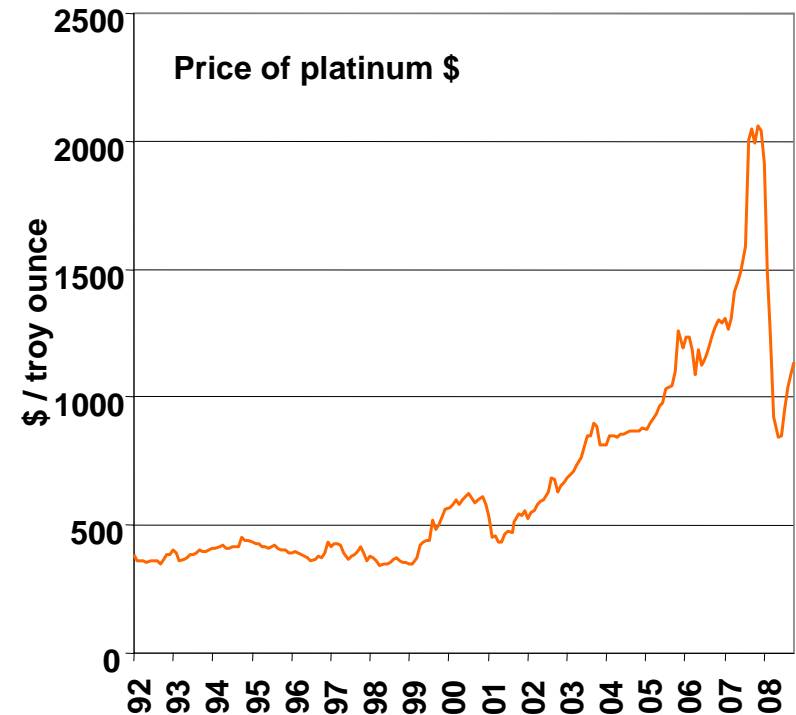
Practical Problems

Start up/shut down
Load Control
Ambient temperature
Durability

GM (May, 2002) Chevrolet S-10 fuel cell demonstration vehicle powered by onboard reformer

Fuel cell as automotive powerplant

- Current (2006) Fuel cell characteristics
 - 1A/cm², 0.5-0.7 V operating voltage
 - 0.5-0.7 W/cm² power density
 - stack power density 0.7 kW/L
 - Platinum loading ~0.3 mg/cm²
 - 30g for a 60kW stack (2007 price ~\$1300)
 - (automotive catalyst has ~2-3g)
 - System efficiency (with reformer) 30%
 - \$600/kW (compared to passenger car at \$10/kW)



Future of Petroleum fueled fuel cell

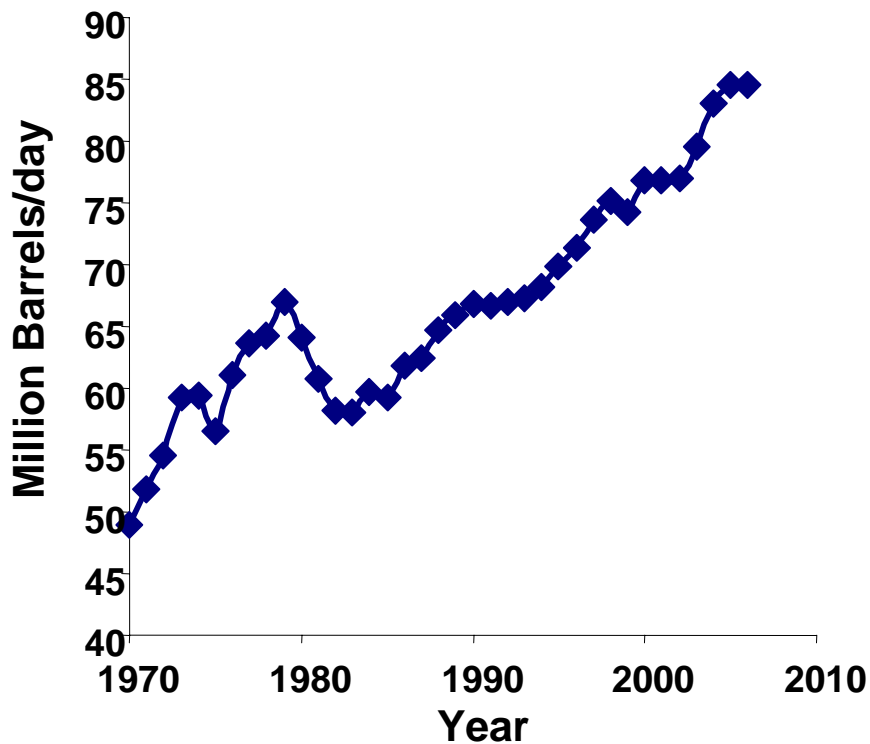
Is the emperor
wearing any
clothes?

- Not an attractive option:
 - Cost
 - Fuel utilization

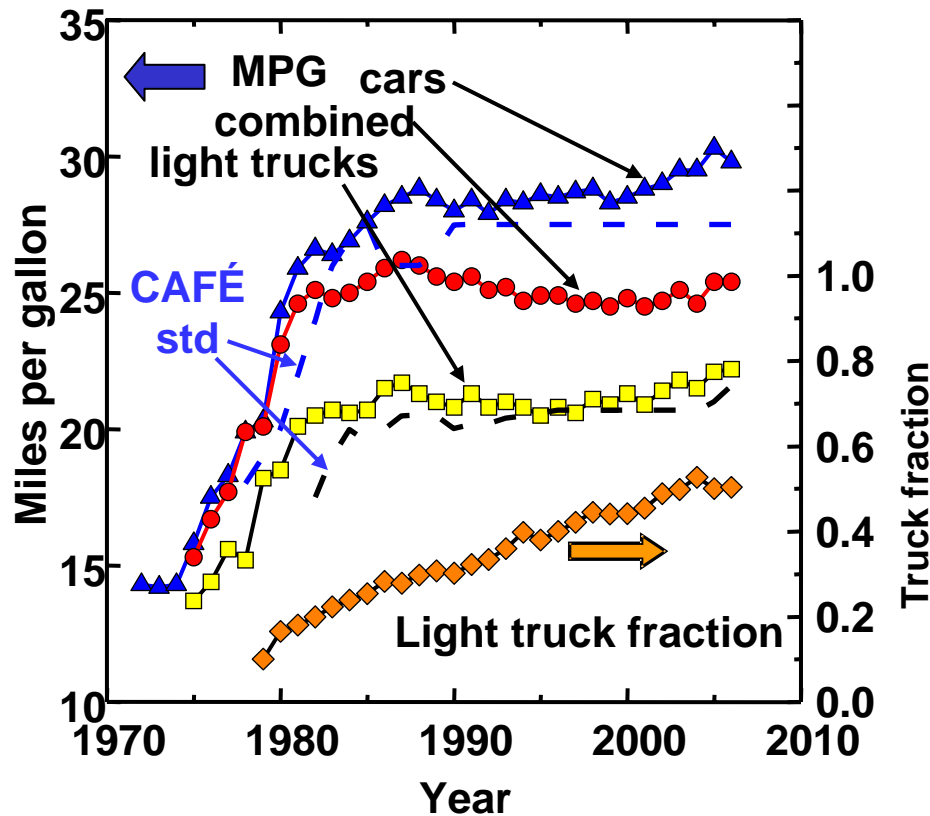
Fuel cell is NOT the technological solution



World Oil Production



US vehicles fuel economy



(Cafe target: 35 mpg average by 2020)

Progress in gas mileage !

Image and text removed due to copyright restrictions. Please see
"Numbers" in *Time Magazine*, June 16, 2003.
<http://www.time.com/time/magazine/article/0,9171,1005048,00.html>

From Time Magazine, June 2003

TRANSPORTATION EFFICIENCY

$$\text{Transportation Efficiency} = \frac{\text{"Useful people mile"}}{\text{Fuel energy}}$$

$$= \frac{\text{"Useful people mile"}}{\text{People mile}} \times \frac{\text{People mile}}{\text{Vehicle mile}} \times \frac{\text{Vehicle mile}}{\text{Road work}} \times \frac{\text{Road work}}{\text{Fuel energy}}$$

Personal efficiency

Vehicle utilization
efficiency

Route, traffic pattern
Vehicle weight/speed

Engineering

Options?

- **Alternative Fuels and Power Plants ?**
- **Alternative Life Styles ?**