## 10.37 Chemical and Biological Reaction Engineering, Spring 2007 Prof. William H. Green Lecture 21: Reaction and Diffusion in Porous Catalyst (cont'd)

This lecture covers: Packed bed reactors





Void Fraction  $\phi \sim .5$ 

$$C_A(x_j, y_m, z_l)$$
  

$$j = 1,30$$
  

$$m = 1,30$$
  

$$l = 1,30$$
  
(points)

(11-21) 
$$D_{i}\nabla^{2}C_{i} - \underline{U} \cdot \nabla C_{i} + r_{i}^{fluid} = 0 \quad \leftarrow \text{ in the fluid } i=1, N_{species}^{fluid}$$
$$D_{i} \frac{\partial C_{i}}{\partial \hat{n}}\Big|_{surface} + r_{i}^{surface} = 0 \qquad (\text{boundary condition for the above})$$

Ergun's Eq.:

(4-22) 
$$\frac{dP}{dz} = -\frac{G}{\rho g_c D_p} \frac{1-\phi}{\phi^3} \left[ \frac{150(1-\phi)\mu}{D_p} + 1.75G \right]$$
  
where  $P = \rho RT$ ,  $\rho U_z = \frac{G}{A}$ 

Cite as: William Green, Jr., course materials for 10.37 Chemical and Biological Reaction Engineering, Spring 2007. MIT OpenCourseWare (http://ocw.mit.edu), Massachusetts Institute of Technology. Downloaded on [DD Month YYYY].

$$\Omega = \frac{\text{Actual rate of reaction } (r_A^{\text{eff}})}{\text{Rate if } C_A = C_{A_{bulk}}(z), \text{ and } T = T_{bulk} \text{ everywhere}}$$

$$\frac{\partial F_i}{\partial z} = Ar_i^{eff} = A\Omega r_i^{ideal} \qquad r_i^{ideal} [=] \frac{\mathsf{mol}}{\mathsf{vol. s}}$$

$$r_i^{ideal} = r_i' \left( \frac{\text{wt. of catalyst in reactor}}{\text{V reactor}} \right)$$

$$r_{i} = r_{i}' \rho_{c} (1-\phi)$$
  
=  $r_{i}'' \left( \frac{\text{surface area of cat.}}{\text{wt. cat.}} \right) \rho_{c} (1-\phi)$   
 $a_{c} / m_{particle}$ 

$$\frac{m^2}{g} = S_a + (\text{macroscopic surface area, visual})$$

$$F_{A} = \nu C_{A} = AUC_{A}$$
 (some approximation)





Figure 2. Flow over a sphere

10.37 Chemical and Biological Reaction Engineering, Spring 2007 Prof. William H. Green

Lecture 21 Page 2 of 3

Cite as: William Green, Jr., course materials for 10.37 Chemical and Biological Reaction Engineering, Spring 2007. MIT OpenCourseWare (http://ocw.mit.edu), Massachusetts Institute of Technology. Downloaded on [DD Month YYYY].

$$\begin{aligned} k_{c}a_{c}\left(C_{A_{b}}-C_{As}\right) &= \eta\left(C_{As}\right)r_{A}^{particle}\left(C_{As}\right)V_{p} \\ D_{inside}\frac{\partial^{2}C_{A}}{\partial r^{2}} + r_{A}\left(C_{A}(r)\right) &= 0 \\ \frac{\partial C_{A}}{\partial r}\Big|_{r=0} &= 0 \\ D_{eff\ inside}\frac{\partial C_{A}}{\partial r} &= k_{c}\left(C_{A_{b}}-C_{As}\right) \\ k_{c}C_{A}\Big|_{r=R} + D_{eff\ inside}\frac{\partial C_{A}}{\partial r}\Big|_{r=R} &= k_{c}C_{A\ bulk} \end{aligned}$$

Matlab:

1) Guess  $C_{As}$  ( $C_A$  surface)

2) Use boundary conditions to get corresponding  $\frac{\partial C_A}{\partial r}\Big|_{r=1}$ 

3) Solve ODE (ode15s)

4) Vary guess (
$$C_{As}$$
) to make  $\frac{\partial C_A}{\partial r} = 0$  at center

1<sup>st</sup> oder irrev.

$$\Omega = \frac{\eta}{1 + \frac{\eta k_1'' S_a \rho_b}{k_c a_c}} \qquad \eta = \frac{3}{\phi_1^2} \left(\phi_1 \operatorname{coth}(\phi_i) - 1\right)$$
$$\phi_1 = \dots \sqrt{k_1''}$$

10.37 Chemical and Biological Reaction Engineering, Spring 2007 Prof. William H. Green